

REACTIONS OF ORGANOMAGNESIUM AND ORGANOLITHIUM COMPOUNDS WITH DIETHYL
(2,3-EPOXYBUTYLIDENE)MALONATE. A SIMPLE SYNTHESIS OF CYCLOPROPANE CARBOXYLATES

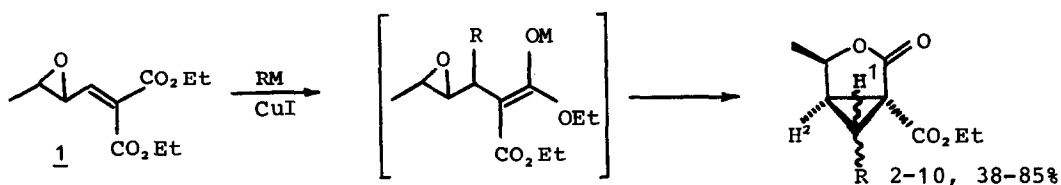
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Summary: Reactions of diethyl (2,3-epoxybutylidene)malonate 1 with various magnesium and lithium organics afford lactones 2-10, each containing a cyclopropane fragment.

Different cyclopropanes containing functional groups can be effectively synthesized via the conjugated addition of nucleophiles to γ -halogen- α , β -unsaturated carbonyl substrates and the subsequent ring-closure of the resulting enolates (MIRC-reactions)¹⁻⁹. Halogen derivatives of alkyl crotonates¹⁻⁵, diethyl alkylidenemalonates⁶⁻⁸, or unsaturated acylphosphoranes⁹ were successfully used in the reactions earlier. This communication is to show diethyl (2-alkenylidene)malonate monoepoxides to be potential substrates for the MIRC-reactions.

On treatment with either a Grignard reagent or a lithium organic compound in ether (-78 \rightarrow 20°C, 3 h) in the presence of 5 mol % CuI, the epoxide 1 has been found¹⁰ to lead to cyclopropane bearing lactones 2-10 (Table 1).



R = Me (2), Pr (3), CH₂=CH (4), Z-i-BuCH=CH (5), 1-cyclohexenyl (6), 1,3-butadien-2-yl (7), Ph (8), PrC \equiv C (9), PhC \equiv C (10); M = Li, MgCl, MgBr, Mgl.

As it follows from the ¹H NMR spectral analysis, the substituent R and the lactone ring are preferably trans-positioned in 2-10. The values of H¹-H²-coupling constants in the isomerically pure cyclopropanes 2, 5, 7, 8 are between 4.5 and 5.1 Hz, while those in compounds 4 and 10 are 4.5 and 8 Hz for the major and minor components.

According to the data given in Table 1, the maximum stereoselectivity is attained with cyclopropanes 2, 5, 7, 8 (minor isomers have not been registered by the ¹H and ¹³C NMR spectroscopy), the ratio E/Z being otherwise varied within 60:40 - 92:8 with the rest.

Table 1. Reactions of Epoxide 1 with Mg and Li Organic Compounds.

Reagent	Product ^c	Yield, % ^a	E/Z ^b
MeMgI	<u>2</u>	65	>95:5
PrMgBr	<u>3</u>	85	60:40
CH ₂ =CHMgBr	<u>4</u>	70	76:24
Z-i-BuCH=CHLi	<u>5</u>	75	>95:5
(1-cyclohexenyl)Li	<u>6</u>	66	92:8
CH ₂ =CHC(MgCl)=CH ₂	<u>7</u>	72	>95:5
PhMgBr	<u>8</u>	69	>95:5
PrC≡CLi	<u>9</u>	38	75:25
PhC≡CLi	<u>10</u>	66	82:18

^athe yields are given for isolated products;

^bby the NMR spectral data.

A typical experimental procedure for the synthesis of lactones 2-10 is as follows. A solution of MeMgI (2.1 mmole) in ether (1 ml) was added to a solution of epoxide 1 (456 mg, 2 mmole) in ether (15 ml) with CuI (19 mg, 0.1 mmole) at -78°C under Ar. The reaction mixture was stirred for 15 min at -78°C, then heated to 20°C for 3 h, and hydrolyzed with 8% HCl (10 ml). A crude product was extracted with ether; the organic layer was dried with MgSO₄ and concentrated. The cyclopropane 2 was isolated as a colorless oil (257 mg, 65%) by column chromatography [silica gel (L 40/100), hexane/ether, 2:1].

m/e 198(M⁺); ¹H NMR (CDCl₃) δ 1.35(3H, t, J= 7.2 Hz), 1.35(3H, d, J= 5 Hz), 1.38(3H, d, J= 5 Hz), 1.8(1H, dq, J₁=J₂= 5 Hz), 2.55(1H, dd, J₁=J₂= 5 Hz), 4.32(2H, q, J= 7.2 Hz), 4.8(1H, dq, J₁=J₂= 5 Hz); ¹³C NMR(CDCl₃) δ 11.92q, 14.30q, 17.50q, 26.33d, 35.98d, 36.46s, 61.82t, 73.96d, 165.67s, 170.67s; IR(film): 1780, 1720 cm⁻¹.

REFERENCES

1. E.Ghera, Y.Ben-David. Tetrahedron Lett., 4603(1979).
2. R.D.Little, J.R.Dawson. Tetrahedron Lett., 21, 2609(1980).
3. P.Prempre, S.Radviroongit, Y.Thebtaranonth. J.Org. Chem., 48, 3553(1983).
4. M.Yamaguchi, M.Tsukamoto, I.Hirao. Tetrahedron Lett., 26, 1723(1985).
5. M.Joucla, M.E.Goumzili, B.Fouchet. Tetrahedron Lett., 27, 1677(1986).
6. P.Kolsaker, H.J.Storesund. J.Chem.Soc.Chem.Comm., 375(1972).
7. S.Torii, H.Tanaka, Y.Nagai. Bull.Chem.Soc.Japan, 50, 2825(1977).
8. M.J.De Vos, A.Krief. Tetrahedron Lett., 1891(1979).
9. M.P.Cook, Jr. and J.Van Jaw. J.Org.Chem., 51, 758(1986).
10. Monoepoxide 1 yielded 63% in reaction of trans-diethyl (2-butenylidene)malonate with monophtalic acid (C= 1.1M) in ether at 20°C.
11. All prepared compounds had satisfactory ¹H and ¹³C NMR and IR spectral assignments; their mass-spectra and elemental analysis data were acceptable too.
12. A.Gordon, P.Ford. Sputnik Khimika, M.: "Mir", 1976, s.301.

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